## 249.

## Problem 26.9 (RHK)

To make some ice, a freezer extracts 185 kJ of heat  $at -12.0^{\circ}C$ . The freezer has a coefficient of performance of 5.70. The room temperature is  $26.0^{\circ}C$ . (a) We have to find the amount of heat that was delivered to the room; and (b) the work required for running the freezer.

## **Solution:**

The amount of heat extracted from the freezer,

 $Q_L = 185 \text{ kJ}.$ 

Coefficient of performance of a refrigerator,

$$K = \frac{Q_L}{W}.$$

As the coefficient of performance of the freezer is 5.70, we find that the mechanical work done on the freezer will be

$$W = \frac{Q_L}{5.70} = \frac{185}{5.7} \text{ kJ} = 32.5 \text{ kJ}.$$

Therefore, the amount of heat delivered to the room will be  $Q_H = Q_L + W = (185 + 32.5) \text{ kJ} = 217.5 \text{ kJ}.$