895.

Problem 55.53 (RHK)

Suppose we had a quantity of N deuterons (²H nuclei). (a) We have to find which one of the following procedures for fusing these N nuclei releases more energy, and how much more. (A) N/2 fusion reactions of the type ${}^{2}\text{H} + {}^{2}\text{H} \rightarrow {}^{3}\text{H} + {}^{1}\text{H}$ (Q = +4.03 MeV), or (B) N/3 fusion reactions of the type ${}^{2}\text{H} + {}^{3}\text{H} \rightarrow {}^{4}\text{He} + n$ (Q = +17.59 MeV), using N/3 nuclei of ${}^{3}\text{H}$ that are first made in N/3 reactions of the type A. (b) We have to list the ultimate product nuclei resulting from the two procedures and the quantity of each.

Solution:

(a)

In N/2 fusion reactions of the type (A)

 $^{2}H + ^{2}H \rightarrow ^{3}H + ^{1}H \qquad (Q = +4.03 \text{ MeV})$

the total energy released will be

$$=\frac{4.03N}{2}$$
 MeV = 2.015N MeV.

We consider next the process of type (B). In N/3 reactions of the type A the energy released will be

$$=\frac{4.03N}{3}$$
 MeV = 1.343N MeV,

and 2*N*/3 deuteron nuclei ²H would have fused and *N*/3 nuclei of ³H would have formed. These *N*/3 nuclei of ³H fuse with the remaining *N*/3 nuclei of ²H by the process ²H + ³H \rightarrow ⁴He + n (*Q* = +17.59 MeV), and released additional energy

$$=\frac{17.59N}{3}$$
 MeV = 5.863N MeV.

Thus the total energy released in the fusion of N

$$=(1.343N + 5.863N)$$
 MeV $= 7.206N$ MeV.

We note that when N nuclei ²H fuse with the process (B) , excess energy of amount 5.191N MeV would be released.

(b)

The ultimate products of fusion process of the type (A) will be N/2 nuclei of ³H, and N/2 protons. And, the ultimate products of fusion process of the type (B) will be N/3 nuclei of ⁴He, and N/3 protons and neutrons each.

