

875.

Problem 55.28 (RHK)

Among the many fission products that may be extracted chemically from the spent fuel of a nuclear power reactor is ^{90}Sr ($t_{1/2} = 29 \text{ y}$). It is produced in typical large reactors at the rate of 18 kg y^{-1} . By its radioactivity it generates thermal energy at the rate of 2.3 W/g . (a) We have to calculate the effective disintegration energy Q_{eff} associated with the decay of a ^{90}Sr nucleus. (Q_{eff} includes contributions from the decay of ^{90}Sr daughter products in its decay chain but not from neutrinos, which escape totally from the sample.) (b) It is desired to construct a power source generating 150 W (electric) to use in operating electronic equipment in an underwater acoustic beacon. If the source is based on the thermal energy generated by ^{90}Sr and if the efficiency of the thermal-electric conversion process is 5.0% , we have to calculate the amount of ^{90}Sr that will be needed. The atomic mass of ^{90}Sr is 89.9 u .

Solution:

(a)

The atomic mass of ^{90}Sr is 89.9 u. Therefore, number of ^{90}Sr atoms contained in 1.0 g of ^{90}Sr will be

$$N = \frac{1.0 \text{ g}}{89.9 \times 1.6605 \times 10^{-24} \text{ g}} = 6.698 \times 10^{21}.$$

As the half-life of disintegration of ^{90}Sr is $t_{1/2} = 29 \text{ y}$, its disintegration constant will be

$$\lambda = \frac{\ln 2}{29 \times 3.156 \times 10^7 \text{ s}} = 7.573 \times 10^{-10} \text{ s}^{-1}.$$

One gram of ^{90}Sr will decay at the rate

$$\begin{aligned} R = \lambda N &= 7.573 \times 10^{-10} \times 6.698 \times 10^{21} \text{ s}^{-1} \\ &= 50.72 \times 10^{11} \text{ s}^{-1}. \end{aligned}$$

By its radioactivity 1.0 g of ^{90}Sr generates thermal energy at the rate of 2.3 J per second. Therefore, the effective disintegration energy Q_{eff} associated with the decay of a ^{90}Sr nucleus will be

$$Q_{\text{eff}} R = 2.3 \text{ J s}^{-1},$$

or

$$\begin{aligned} Q_{\text{eff}} &= \frac{2.3 \text{ J s}^{-1}}{50.72 \times 10^{11} \text{ s}^{-1}} = 4.53 \times 10^{-13} \text{ J} \\ &= \frac{4.53 \times 10^{-13}}{1.6 \times 10^{-13}} \text{ MeV} = 2.8 \text{ MeV}. \end{aligned}$$

(b)

It is desired to construct using ^{90}Sr a power source generating 150 W (electric) to use in operating electronic equipment in an underwater acoustic beacon. It is given that the efficiency of the thermal-electric conversion process is 5.0%. Therefore, the amount of ^{90}Sr required will be

$$2.3 \times 0.05 \times m \text{ W g}^{-1} = 150 \text{ W},$$

or

$$m = \frac{150}{2.3 \times 0.05} \text{ g} = 1.3 \text{ kg}.$$

