

804.

**Problem 53.27 (RHK)**

*Silver is a monovalent metal. We have to calculate (a) the number of conduction electrons per cubic meter, (b) the Fermi energy, (c) the Fermi speed, and (d) de Broglie wavelength corresponding to this speed.*

**Solution:**

(a)

From the physical data for silver, we note that molar mass of silver,  $M_{\text{Ag}} = 107.68 \text{ g mol}^{-1}$ , density of silver,  $\rho_{\text{Ag}} = 10.49 \text{ g cm}^{-3}$ .

Therefore, the number of silver atoms per cubic meter will be

$$n = \frac{N_A \rho_{\text{Ag}}}{M_{\text{Ag}}} = \frac{6.02 \times 10^{23} \text{ mol}^{-1} \times 10.49 \times 10^3 \text{ kg m}^{-3}}{107.68 \times 10^{-3} \text{ kg mol}} \\ = 5.86 \times 10^{28} \text{ m}^{-3}.$$

As silver is a monovalent metal, each atom of silver contributes one conduction electron. Therefore, the number of conduction electrons per cubic meter of silver

will be equal to the number of silver atoms per cubic meter in metallic silver.

(b)

We calculate next the Fermi energy for the conduction electrons in metallic silver using the relation

$$E_F = \frac{h^2}{8m} \left( \frac{3n}{\pi} \right)^{2/3}.$$

Using that for silver  $n = 5.86 \times 10^{28} \text{ m}^{-3}$ , we find that for conduction electrons of silver the Fermi energy will be

$$\begin{aligned} E_F &= \frac{h^2}{8m} \left( \frac{3n}{\pi} \right)^{2/3} = \frac{(6.63 \times 10^{-34})^2}{8 \times 9.11 \times 10^{-31}} \times \left( \frac{3 \times 5.86 \times 10^{28}}{\pi} \right)^{2/3} \text{ J} \\ &= 8.822 \times 10^{-19} \text{ J} \\ &= 8.822 \times 10^{-19} \times 6.242 \times 10^{18} \text{ eV} \\ &= 5.51 \text{ eV}. \end{aligned}$$

(c)

We calculate next the Fermi speed for conduction electrons in silver metal. We have

$$\frac{1}{2} m v_F^2 = E_F,$$

or

$$\begin{aligned} v_F &= \left( \frac{2E_F}{m} \right)^{1/2} = \left( \frac{2 \times 8.822 \times 10^{-19} \text{ J}}{9.11 \times 10^{-31} \text{ kg}} \right)^{1/2} \\ &= 1.39 \times 10^6 \text{ m s}^{-1}. \end{aligned}$$

(d)

The de Broglie wavelength of conduction electrons moving with Fermi speed  $v_F$  will be

$$\lambda_F = \frac{h}{mv_F} = \frac{6.63 \times 10^{-34}}{9.11 \times 10^{-31} \times 1.39 \times 10^6} \text{ m}$$
$$= 5.236 \times 10^{-10} \text{ m} = 523.6 \text{ pm.}$$

