

764.

**Problem 40.51P (HRW)**

*The wave functions for the three states of hydrogen atom, which have  $n=2$ ,  $l=1$ ,  $m_l=0$ ,  $+1$ , and  $-1$ , are*

$$\psi_{210}(r, \theta) = \left(1/4\sqrt{2\pi}\right) \left(a^{-3/2}\right) (r/a) e^{-r/2a} \cos \theta,$$

$$\psi_{21+1}(r, \theta, \phi) = \left(1/8\sqrt{\pi}\right) \left(a^{-3/2}\right) (r/a) e^{-r/2a} \sin \theta e^{i\phi},$$

$$\psi_{21-1}(r, \theta, \phi) = \left(1/8\sqrt{\pi}\right) \left(a^{-3/2}\right) (r/a) e^{-r/2a} \sin \theta e^{-i\phi},$$

*in which the subscripts on  $\psi(r, \theta, \phi)$  give the values of the quantum numbers  $n$ ,  $l$ ,  $m_l$ . Note that the first wave function is real but the others, which involve the imaginary number  $i$ , are complex.*

*We have to find the probability density for each function and by adding the three probability densities show that their sum is spherically symmetric, depending on the radial coordinate  $r$ .*

**Solution:**

(a)

The probability density for a quantum state is given by the modulus square of the normalized wave function.

The probability densities of the states corresponding to the quantum numbers,  $n = 2$ ,  $l = 1$ ,  $m_l = 0$ ,  $+1$ , and  $-1$ , will therefore be

$$P_{210}(r, \theta, \phi) = |\psi_{210}(r, \theta, \phi)|^2 = \frac{1}{32\pi} \frac{1}{a^5} r^2 e^{-r/a} \cos^2 \theta,$$

$$P_{211}(r, \theta, \phi) = |\psi_{211}(r, \theta, \phi)|^2 = \frac{1}{64\pi} \frac{1}{a^5} r^2 e^{-r/a} \sin^2 \theta,$$

and

$$P_{21-1}(r, \theta, \phi) = |\psi_{21-1}(r, \theta, \phi)|^2 = \frac{1}{64\pi} \frac{1}{a^5} r^2 e^{-r/a} \sin^2 \theta.$$

Therefore, we get

$$\begin{aligned} & |\psi_{210}(r, \theta, \phi)|^2 + |\psi_{211}(r, \theta, \phi)|^2 + |\psi_{21-1}(r, \theta, \phi)|^2 \\ &= \frac{1}{32\pi} \frac{1}{a^5} r^2 e^{-r/a} \cos^2 \theta + \frac{1}{32\pi} \frac{1}{a^5} r^2 e^{-r/a} \sin^2 \theta \\ &= \frac{1}{32\pi} \frac{1}{a^5} r^2 e^{-r/a}. \end{aligned}$$