

760.

Problem 40.49P (HRW)

We have to show that the average value of r , defined as

$$\bar{r} = \int_0^{\infty} rP(r)dr,$$

has the value $1.5a$. $P(r)$ is the radial probability density for the ground state of the hydrogen atom.

Solution:

The radial probability density for the ground state of the hydrogen atom is



$$P(r) = \frac{4}{a^3} e^{-2r/a},$$

where a is the Bohr radius.

Therefore,

$$\bar{r} = \int_0^{\infty} rP(r)dr = \frac{4}{a^3} \int_0^{\infty} r^3 e^{-2r/a} dr.$$

We evaluate the integral by making the following substitutions:

$$\frac{2r}{a} = \xi, \quad dr = \frac{a}{2} d\xi.$$

We get

$$\bar{r} = \frac{4}{a^3} \left(\frac{a}{2} \right)^4 \int_0^\infty \xi^3 e^{-\xi} d\xi = \frac{a}{4} \Gamma(4) = \frac{a}{4} 3! = \frac{6a}{4} = 1.5a.$$

