709.

Problem 49.18 (RHK)

Given that an ideal radiator has a spectral radiancy at 400 nm that is 3.50 times its spectral radiancy at 200 nm. (a) We have to find the temperature of the radiator. (b) We have to calculate the temperature if the spectral radiancy of the radiator at 200 nm were 3.5 times its spectral radiancy at 400 nm.

Solution:

According to Planck's radiation law the spectral radiancy of a radiator at temperature T and wavelength λ is given by the following function:

$$R(\lambda) = \frac{2\pi c^2 h}{\lambda^5} \times \frac{1}{\exp(hc/\lambda kT) - 1},$$

where

the Boltzmann constant $k = 1.381 \times 10^{-23} \text{ J K}^{-1}$ and the Planck constant $h = 6.626 \times 10^{-34} \text{ J s}$ (a)

It is given that at temperature *T* the radiancy at $\lambda = 200$ nm and at $\lambda' = 2\lambda = 400$ nm are related as follows:

$$R(\lambda' = 400 \text{ nm}) = 3.5 \times R(\lambda = 200 \text{ nm}).$$

We therefore have the equation

$$\frac{1}{\left(\lambda'\right)^{5}} \times \frac{1}{\exp\left(hc/\lambda'kT\right) - 1} = 3.5 \times \frac{1}{\lambda^{5}} \frac{1}{\exp\left(hc/\lambda kT\right) - 1}$$

or

$$\frac{1}{2^5} \times \frac{1}{x-1} = 3.5 \times \frac{1}{x^2 - 1}$$

where
 $x = \exp(hc/2\lambda kT)$.

The above equation readily simplifies to the form

$$x^{2} - 112x + 111 = 0,$$

or
 $(x-1)(x-111) = 0.$

Root x = 1 is unphysical. We find temperature *T* corresponding to the root

$$x = 111,$$

we have
$$\frac{hc}{2\lambda kT} = \ln(111),$$

or
$$T = \frac{6.626 \times 10^{-34} \times 3 \times 10^{8}}{400 \times 10^{-9} \times 1.381 \times 10^{-23} \times \ln(111)} \text{ K}$$
$$= 7641 \text{ K}.$$

(b)

We next solve the second part of the problem.

The condition is

$$R(200 \text{ nm}) = 3.5 \times R(400)$$

We once again make the

$$x = \exp(hc/2\lambda kT).$$

For this case, we solve the following equation for finding the temperature of the radiator:

$$\frac{1}{x^2 - 1} = \frac{3.5}{2^5} \times \frac{1}{x - 1},$$

or

$$(x-1)\left(x-\left(\frac{2^5}{3.5}-1\right)\right)=0.$$

We select the physical root

$$x = \left(\frac{2^5}{3.5} - 1\right) = 8.143.$$

We next solve

 $\exp(hc/2\lambda kT) = 8.143,$

for

 $\lambda = 200 \text{ nm} = 2.0 \times 10^{-7} \text{ m},$ and find that

T = 17,170 K.

