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Problem 32.35 (RHK)

Copper and aluminium are being considered for a high-voltage transmission line that must carry a current of 62.3 A. The resistance per unit length is to be $0.152 \Omega \text{ km}^{-1}$. We have to compute for each choice of material (a) the current density and (b) the mass of 1.00 m of cable. The densities of copper and aluminium are 8960 and 2700 kg m^{-3} , respectively.



Solution:

The mass densities of copper and aluminium are

$$\rho_{m,Cu} = 8960 \text{ kg m}^{-3},$$

and

$$\rho_{m,Al} = 2700 \text{ kg m}^{-3}.$$

And electrical resistivities of these materials are

$$\rho_{c,Cu} = 1.69 \times 10^{-8} (\Omega \text{ m}),$$

and

$$\rho_{c,Al} = 2.75 \times 10^{-8} (\Omega \text{ m}).$$

(a)

We will first find the cross-sectional areas of the copper and aluminium wires, having resistance per unit length of $0.152 \Omega \text{ km}^{-1}$. We will use the result that resistance, R , of a wire of length, l , resistivity, ρ_c , and cross-sectional area, A , is given by

$$R = \frac{\rho_c l}{A}.$$

Therefore, for copper wire

$$A_{cu} = \frac{\rho_{c,Cu} \times l}{R} = \frac{1.69 \times 10^{-8} \times 1000}{0.152} \text{ m}^2 = 1.118 \times 10^{-4} \text{ m}^2 \\ = 1.118 \text{ cm}^2.$$

And, similarly for aluminium wire

$$A_{Al} = \frac{\rho_{c,Al} \times l}{R} = \frac{2.75 \times 10^{-8} \times 10^3}{0.152} \text{ m}^2 = 1.809 \times 10^{-4} \text{ m}^2 \\ = 1.809 \text{ cm}^2.$$

The transmission line is to carry current of 62.3 A.

Therefore, the current density in the copper cable will be

$$j_{Cu} = \frac{62.3}{1.118 \times 10^{-4}} \text{ A m}^{-2} = 5.57 \times 10^5 \text{ A m}^{-2} = 55.7 \text{ A cm}^2,$$

and the current density in the aluminium cable will be

$$j_{Al} = \frac{62.3}{1.809 \times 10^{-4}} \text{ A m}^{-2} = 3.44 \times 10^5 \text{ A m}^{-2} = 34.4 \text{ A cm}^2.$$

(b)

Mass of 1-m of copper cable will be

$$m_{Cu} = \rho_{m,Cu} \times A_{Cu} = 8960 \times 1.118 \times 10^{-4} \text{ kg} = 1.001 \text{ kg},$$

and the mass of 1-m of aluminium cable will be

$$m_{Al} = \rho_{m,Al} A_{Al} = 2700 \times 1.809 \times 10^{-4} \text{ kg} = 0.488 \text{ kg}.$$

