## **Problem 27.26 (RHK)**

The electrostatic force between two identical ions that are separated by a distance of  $5.0 \times 10^{-10}$  m is  $3.7 \times 10^{-9}$  N. (a) We have to find the charge on each ion. (b) We have to find the number of electrons that are missing from each ion.

## **Solution:**

Let the charge on each ion be q C. Separation of the two ions is

$$r = 5.0 \times 10^{-10}$$
 m.

The magnitude of electrostatic force on each ion due to the other is

$$F = 3.7 \times 10^{-9} \text{ N}.$$

Therefore, by Coulomb's law we have

$$\frac{q^2}{4\pi\varepsilon_0 r^2} = F.$$

And

$$q = \left(\frac{3.7 \times 10^{-9} \times 25 \times 10^{-20}}{8.99 \times 10^{9}}\right)^{\frac{1}{2}} \text{ C} = 3.2 \times 10^{-19} \text{ C}.$$

The magnitude of charge on an electron is

$$e = 1.6 \times 10^{-19}$$
 C.

Therefore,

$$q = 2e$$
.

We thus find that 2 electrons are missing from each ion for it to have 2 units of positive charge.

